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Multiple Choice
Questions 1) When

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Nuclear Chemistry

Section 22.1

more than 3600 known nuclides are plotted on a neutron/proton grid they make up a group called A) the "island of stability." B) the "peninsula of nuclear stability." C) the "sea of instability."

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Nuclear Chemistry-

Chapter 22. STUDY.

PLAY. nucleons.

description of atomic

nuclei based on proton

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Section 22.1 Answers

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and neutrons particles.
nuclide. used in
nuclear chemistry as a
reference to an atom,
identified by the
number of protons and
neutrons in its nucleus
(another name for an
isotope ... Chapter 22
Review: Nuclear
Chemistry. 28 terms.
common ions (name ...

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Nuclear Chemistry.

Section 22.1
History Just because it
is interesting. Marie

Curie: Henri Becquerel
(Physics professor,
1852 - 1908)

discovered
radioactivity in 1896 by
placing a photographic
plate near a box of
uranium ore. When the
plate "developed" on
its own, he knew the
ore contained
something giving off
energy.

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Nuclear Chemistry

Chapter 22 Review **Nuclear Chemistry**

22Ne?? Nuclear
Reactions Unstable
nuclei undergo
spontaneous changes
that change their num-
ber of protons and
neutrons. In this
process, they give off
large amounts of
energy and increase
their stability. These
changes are a type of
nuclear reaction. A
nuclear reaction is a
reaction that affects

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the nucleus of an atom.

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of atom or nucleus
characterized by a
specific number of
protons and neutrons
Mass Defect difference
between the mass of
an isotope and its mass
number Nuclear
Binding energy is the
energy that would be
required to
disassemble the
nucleus of an atom into
its component part
which would be
neutrons [...]

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SECTION 22-4 SHORT

ANSWER Answer the following questions in the space provided. 1.

Label each of the following statements with one of the choices below: (1) fission only (3) both fission and fusion

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152 Nuclear chemistry

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Section 22.1 Answers

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involves changes in the nucleus (protons and neutrons) of radioactive atoms.

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treatment C-14 dating

nuclear power plants

create new elements

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Nuclei and Nuclear

Reactions

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Number Textbook:

Chemistry and

Chemical Reactivity

(4th Edition) by Kotz

and Treichel Textbook:

ISBN 0030237629 /

Study Guide: ISBN

0030237947. eXam

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your answers to the
questions and click
'Next' to see the next

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spontaneous
disintegration of a
nucleus into a slightly
lighter and more stable
nucleus, accompanied
by emission of
particles,
electromagnetic
radiation, or both ...

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Transformations •
Rutherford in 1919

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Section 22.1 Transmutations

Modern

performed the first nuclear transformation.

- The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes: ${}_{14}^{27}\text{Al} + {}_2^4\text{He} \rightarrow {}_{17}^{31}\text{P} + {}_1^1\text{H}$

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Nuclearchemistry

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SECTION 2 SHORT
ANSWER Answer the
following questions in
the space provided. 1.

_____ The nuclear
equation is an example
of an equation that
represents (a) alpha
emission. (b) beta
emission. (c) positron
emission. (d) electron
capture. 2.

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02ec675.