

Modern Chemistry Chapter Test B Chemical Bonding

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued ____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. ____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n). 18. The decomposition of a substance by an electric current is called.

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Modern Chemistry 132 Chapter Test Name Class Date Chapter Test B, continued Refer to the following statement to answer questions 30-32: Dilute HCl(aq) and NaOH(aq) are mixed in chemically equivalent quantities. 30. Write the chemical equation for the reaction. 31. Write the overall ionic equation for the reaction. 32. Write the net ionic equation.

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Chapter 1 Practice Test B Chapter: Matter and Change ____ 1. Chemistry is a natural science that deals with the study of a. living things and their life processes. b. the physical features of Earth. c. the composition, structure, properties, and changes of matter. d. the composition, motion, and relative positions of stars and planets. ____ 2.

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Holt Chemistry 40 Matter and Energy Section: Measurements and Calculations in Chemistry In the space provided, write the letter of the term or phrase that best answers the question. ____ 1. A measurement is said to have good precision if it a. agrees closely with an accepted standard. b. agrees closely with other measurements of the same quantity.

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b. Explain how an atom can exist in this state. Atoms consist of a positively charged nucleus, made up of protons and neutrons, that is surrounded by a negatively charged electron cloud. The positive and negative charges combine to form a net neutral charge. MODERN CHEMISTRY ATOMS: THE BUILDING BLOCKS OF MATTER 19

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

[Chapter 1 Practice Test B | Chemical Elements | Chemical ...](#)

Modern Chemistry 11 . Name Section Quiz, continued Class Date 6. A measurement is accurate if it a. is reproducible. b. close to the true value. c. has many decimal places. d. has many significant figures. 7. To determine the number of significant digits in a measurement, follow

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Modern Chemistry 111 Chapter Test Name Class Date Chapter Test B, continued 18. Dissolution processes with negative enthalpies of solution are processes. 19. is the solution process with water as the solvent. 20. Liquid solutes and solvents that are not soluble in each other are. 21.

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CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9. If a particular ompound is composed of elements A and B, the ratio of the mass of B to the mass of A will alw ys be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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