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Formulas (pages 170–175) This section explains how to name and write formulas for ionic and molecular compounds. Reading Strategy (page 170)

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Section 6.3 Naming Compounds and
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Matter has two kinds of
properties—physical and chemical.
Freezing point is a physical prop-erty

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and ability to combine with oxygen is a chemical property. The melting of ice to water is an example of a physical change, and the burning of wood is an example of a chemical change. A mixture is made of two or more substances. A

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____ 6. What is one way to increase the solubility of sugar in water? a. Heat the water. b. Chill the water. c. Increase the amount of sugar. d. Decrease the amount of water. ____ 7. Which is a likely use for a base? a. as a vitamin in your food b. etching metals for printing c. making foods taste sour d. making soaps and detergents ____ 8.

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