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... ground state, (i) excited state, (j)
atomic mass, (k) quantum ... Dalton had
the first scientifically supported atomic
theory.

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the answer. The startling ideas of
quantum theory are the basis for the ...
this instrument made to the study of
matter at the atomic level, ... The basis
of the STM can be explained by quantum
theory, but the STM is not the only
phenomenon with roots in quantum the-
ory. Although the theory explains
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Chapter 27 Quantum and Relativistic Physics 595! the radiation emitted from the object and absorbed from the chamber, $I = \epsilon\sigma (T_1^4 - T_2^4)$ (27.4) Wien's law is an empirical equation for the relationship between the temperature of a perfect radiator and the wavelength of the radiation of maximum intensity. It can be

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Pacing Guide 1 period Lesson and Problem-Solving Lab Quantum Theory and the Atom pages 127–134 BLOCK SCHEDULE LESSON PLAN 5.2 Objectives

- Compare the Bohr and quantum mechanical models of the atom.
- Explain the impact of de Broglie's wave-particle duality and the Heisenberg

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uncertainty principle on the modern view of electrons in atoms.

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following questions regarding the discovery of the atom & its parts. A. Identify 3 conclusions of the gold foil experiment. 1-Atom was more than empty space. 2-There is a dense portion in the center of the atom that contained most of its mass. 3-The nucleus must be positively charged. B.

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